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Yttria-stabilized zirconia ($xY_{2O_3} \cdot (1-x)ZrO_2$, YSZ) has been exploited as a fast oxygen-ion conductor at elevated temperature, which is benefited from the formation of oxygen vacancies upon the stabilization of zirconia by yttria doping. Oxygen ion conduction in YSZ takes place by exchanging oxygen ions and vacancies within the cubic/tetragonal crystal framework, therefore, the coordination chemistry of a Zr ion by oxygen are of key importance in terms of crystal structure and crystal field to understand the origin of fast ionic conduction in YSZ. Herein, the Zr-O metal-ligand interactions are modulated by heterostructuring of YSZ with Sm-doped ceria, by which the vibrational properties of Zr-O bonding are altered. The coordination chemistry of the heterostructures is studied by X-ray absorption spectroscopy on Zr L edges, which is combined with neutron diffraction and inelastic neutron scattering studies. The coordination chemistry model for the heterostructures is suggested in conjunction with ab initio calculations.