



LEACHING OF SYNROC IN D2O

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The use of D_2O rather than H_2O for leaching Synroc and its constituent phases allows the study of hydration and hydroxylation reactions without the contaminating influence of atmospheric moisture. For Synroc leached at 190°C for 5-30 days, the penetration of D into the Synroc is found as ~30 nm from both energy recoil analysis using 1.5 MeV He ions from a Van de Graaff accelerator and secondary ion mass spectroscopy near mass 18(OD-).

Analogous studies have been carried out on single crystals of SrTiO₃ which have the perovskite (a constituent phase of Synroc) structure, and augmented by additional atomic force and scanning electron microscope studies of secondary TiO₂ alteration products. Alteration products on single crystals of CaTiO₃ have also been studied.