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AUSTRALIAN ATOMIC ENERGY COMMISSION
RESEARCH ESTABLISHMENT
LUCAS HEIGHTS

ADLI - AN ATOMIC DATA LIBRARY FOR USE IN COMPUTING THE
BEHAVIOUR OF PLASMA DEVICES INCLUDING FUSION REACTORS

by

B.E. CLANCY
J.L. COOK
E.K. ROSE

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ADL1 - AN ATOMIC DATA LIBRARY FOR USE IN COMPUTING THE
BEHAVIOUR OF PLASMA DEVICES INCLUDING FUSION REACTORS

by

B.E. CLANCY
J.L. COOK
E.K. ROSE

ABSTRACT

A data library with self-descriptive format is presented. This library provides, on a fixed temperature grid, reaction rate coefficients, effective degree of ionisation, and data for line radiation power emission for 59 ion or neutral species. Data are presented for neutral and ionised atoms of the hydrogen isotopes and for 49 'impurity' ion species ranging from helium-3 and -4 to uranium. Data origins are also discussed.

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The following descriptors have been selected from the INIS Thesaurus to describe the subject content of this report for information retrieval purposes. For further details please refer to IAEA-INIS-12 (INIS: Manual for Indexing) and IAEA-INIS-13 (INIS: Thesaurus) published in Vienna by the International Atomic Energy Agency.

NUCLEAR DATA COLLECTIONS; IMPURITIES; PLASMA; IONIZED GASES; HYDROGEN;
DEUTERIUM; TRITIUM; TOKAMAK DEVICES; ELECTRICAL PROPERTIES; IONIZATION;
POWER COEFFICIENT; PLASMA DIAGNOSTICS

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1. INTRODUCTION

Several years ago, the development of a computer code SCORCH to simulate the behaviour of tokamak devices was begun. This code uses the plasma transport theory described by Hinton and Hazeltine [1976] and computes the time development of various plasma properties, such as the average number densities, temperatures and the associated losses and gains. Parallel to this development, a preliminary data library was compiled by the present authors to enable them to ascertain the importance of impurity species in calculating the properties of the plasma. Most of the data for the impurities treated were derived from a study of the systematics of the corona model reported in a review article by Drawin [1977], while information on ionisation and recombination rates was taken from a review by McWhirter [1977].

It is clear that compilation of data for impurities in all states of ionisation/excitation is impracticable, since detailed data are not obtainable, and even if they were, this would lead to a data library of prohibitive size. The library therefore contains data for both ionised and neutral atoms of the hydrogen isotopes H, D and T; atoms with a nuclear charge greater than one are compiled as if they had a degree of ionisation $\langle Z \rangle$ equal to that expected for specified electron temperatures in the plasma. The expected value $\langle Z \rangle$ is one of the properties compiled in the library. Partly for reasons which reflect the development history of SCORCH, data for neutral atoms of light hydrogen are compiled in two forms. In one of these, neutral hydrogen may be treated as consisting of the first four states, H_{1s} , H_{2s} , H_{2p} , H_{3s} ; in the second, neutral hydrogen consists simply of a species named HO. If the first form is used, power radiated from individual lines could be evaluated for use in diagnostics.

Data are compiled for many ion species which will not be present in any plasma systems but they have been included to allow for later study of systematics of the power radiation coefficients.

2. LIBRARY FORMAT

The format chosen for the library reflects the dual design requirements that (a) it should be easy to list and modify with the text editors available at the AAEC Research Establishment, and (b) it should be open-ended — in the sense that addition of further materials to the library should be simple. The

library is thus comprised of card images and two main sections. The second section contains a set of tables, each of which has the data values for a particular property of a particular atom/ion, while the first section takes the form of a set of indexes, many of which spell out, for a given property, the particular table number which contains the appropriate data. Each index is preceded by two cards which describe the index entries.

3. LIBRARY DETAILS

3.1 Indexes to the Tables

After a library title card, the first index appears preceded by two header records. Then follows one card for each atom/ion species referenced in the library; this card contains the species name (up to 8 characters), followed by the mass number and nominal charge carried by the species. For the specifically neutral species, this charge number is zero. To signal the end of the species list, a separator card beginning ENDLIST is inserted.

Index 2 then appears, again preceded by two header cards. This index identifies the tables which contain the average degree of ionisation $\langle Z \rangle$ for individual species; each card in the body of the index contains the species name and the appropriate table number. The index ends with a separator card.

The body of index 3 simply contains decay constants λ_{ij} in s^{-1} for the specific set of neutral hydrogen species, H_{1s} , H_{2s} , etc. The decay constants are defined so that if the only reactions occurring were these decays, the number densities of the species would be related by the equation

$$\frac{dn_j}{dt} = \sum_i \lambda_{ij} n_i - \left(\sum_i \lambda_{ji} n_j \right) \cdot$$

The body of index 4 contains data for the energy appearing from the fusion reactions. For each process referenced, a single card appears containing the names of the two initiating ions, the names of the two reaction products, and the energy (J) appearing with each product ion. The end of the list is marked by a separator.

Index 5 specifies tables which contain power density coefficients P_i for individual species. Thus, if number densities for electrons and a species are n_e , n_i (m^{-3}), the power density radiated is $P_i n_e n_i$ ($W m^{-3}$). The individual

cards of index 5 contain a species name and the table number in which P_i is tabulated against electron temperature. This radiated power comes from dielectronic and radiative recombination, electron collisional ionisation and excitation and the inverse processes, and radiative line transitions. Bremsstrahlung energy is specifically omitted. A separator card ends this index.

Index 6 specifies the tables in which are compiled reaction rate coefficients for elastic scattering - usually momentum transfer through Coulomb effects. Thus, if two species are present with number densities n_i , n_j (m^{-3}), the collision rate ($m^{-3} s^{-1}$) between these species is $C_{ij} n_i n_j$ or $C_{ji} n_i n_j$. In the index, one card appears for each pair i, j for which data have been compiled. This card has the names of the two species and the table number in which the coefficient C_{ij} is tabulated against the temperature of the first named species. Electron-ion collisions are included in this index. A separator card closes index 6.

Index 7 specifies tables for rate coefficients of those reactions in which the emerging species are not identical to the species initiating the reaction. Thus, if species i, j react and species k, l are emitted as the reaction products, two rate coefficients will be compiled, C_{ijk} and C_{ijl} . These are defined so that if the number densities of species i, j are n_i , n_j , then the rate ($m^{-3} s^{-1}$) at which particles of species k emerge from this reaction per unit volume is $n_i n_j C_{ijk}$. For such reactions as ionisation by proton collision, species i and k may be identical. Each card in the index contains the names of species i, j, k and the table number in which C_{ijk} is tabulated as a function of the temperature of species i . As with the elastic collision rate coefficients, species i may be an electron. In both index 6 and index 7, the rate coefficients are compiled so that if species i, j are identical, the reaction rate density will be $\frac{1}{2} n_i n_i C_{ii}$ or $\frac{1}{2} n_i n_i C_{iik}$. A separator card signals the end of this final index.

3.2 Tables

The first table specifies the temperature values at which all coefficients are tabulated and consists of four heading cards followed by a set of energy values in keV from which the temperatures (K) can be calculated. The header cards contain the number of energies/temperatures which are used, as well as the conversion factor to convert from keV to K. The use of an energy scale facilitates comparison with other compilations. The present

version of the library uses 45 energy values spanning the range 10 eV to 1 MeV.

The remaining tables have a fixed layout similar to that for the temperature table. The first card of a table contains the table number, and the second card a multiplier which must be applied to the tabulated values to convert them to SI units. Most of the reaction rate coefficients were originally computed in CGS units, and the multiplier is then 10^{-6} . Next come two comment cards which describe the table followed by the tabulated values in fixed format.

4. SOURCES FOR COMPILED DATA

Data for the effective degree of ionisation and for the radiative power coefficients were taken primarily from the calculated results of Post et al. [1977]. Their results are presented in the form of polynomial coefficients applicable over various energy ranges and the irregularities at the energy range break points were smoothed out by eye after subtraction of the bremsstrahlung power, which was also calculated according to Post et al. Power rate coefficients for hydrogen were taken from Drawin [1977] and McWhirter [1977].

The decay constants for hydrogen levels were taken from Weise [1966] and the fusion energy release values from Ribe [1975].

Electron-ion Coulomb scattering was computed from the standard cross section formula reported, for instance, by Glasstone and Lovberg [1950]. These data assume a Coulomb logarithm value $\ln\Lambda = 20$, so calculations using the data must incorporate the correction for $\ln\Lambda$ to agree with the formulae given by Braginski [1965]:

$$\ln\Lambda = 23.4 - 1.15 \log n_e + 3.45 \log T_e ; T_e < 50 \text{ eV}$$

$$\ln\Lambda = 25.3 - 1.15 \log n_e + 2.3 \log T_e ; T_e > 50 \text{ eV} ,$$

where n_e is the electron number density (cm^{-3}) and T_e is the 'temperature' in eV. For partially ionised materials, the $\langle Z \rangle$ value appropriate to the electron temperature was used in the calculation. As ion-ion Coulomb scattering rates were computed according to the same procedure, corrections

for $\alpha n \lambda$ must be applied to them.

Recombination rate coefficients were computed according to a theory of Spitzer [1956]. All other neutral-ion, electron-neutral and neutral-neutral collision rate coefficients were calculated from the cross sections given by Barnett et al. [1977], and the low energy points were taken from Takayanagi and Suzuki [1978]. Ionisation rates were computed from formulae given by McWhirter [1977].

The D-D and D-T fusion rate coefficients were computed from the formulae quoted by Glasstone and Lovberg [1950].

Since much of the data behaves approximately like a power of the temperature, log-log interpolation should be used for intermediate values.

5. LIBRARY PROCESSING PROGRAMS

The pointer-to-table concept used in the indexes is flexible and makes it easy to insert new material in the library; however, it is even easier to insert a wrong pointer value. When checking the library, we found it essential to have library processing programs to provide a library cross reference, to produce plots of various tables and to modify tables found to be in error.

6. REFERENCES

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APPENDIX A
CURRENT LISTING OF ADL1 INDEX

```

=== SCURCH LIBRARY === BEC.ADL1.FASYREAD =====
SPECIES LIST
ATOM      MASS NO.  CHARGE NO.
H+        1         1
H1S       1         0
H2S       1         0
H2P       1         0
H3S       1         0
HQ        1         0
D+        2         1
DC        2         1
T+        3         1
TC        3         0
HELIUM3   3         2
HELIUM4   4         2
LITHIUM   7         3
BEPYLIUM  9         4
BOFCN    10        5
CARBON    12        6
NITROGEN  14        7
OXYGEN    16        8
FLUORINE  19        9
NEON     20       10
SODIUM    23       11
MAGNESIUM 24       12
ALUMINUM  27       13
SILICON   28       14
SULPHUR   32       16
ARGON     40       18
PGTASSUM  39       19
CALCIUM   40       20
SCANDIUM  45       21
TITANIUM  48       22
VANADIUM  51       23
CHROMIUM  52       24
IRON      56       26
COBALT    59       27
NICKEL    59       28
COPPER    64       29
ZINC      65       30
ARSENIC   75       33
KRYPTON   84       36
RUBIDIUM  86       37
ZIRCONIUM 91       40
NIOBIUM   93       41
MOLYBDEN  96       42
RHODIUM   103      45
SILVER    108      47
TIN       119      50
XENON     131      54
CESIUM    133      55
BARIUM    137      56
GADOLINIUM 157     64
TANTALUM  181     73
TUNGSTEN  184     74
IRIDIUM   192     77
GOLD      197     79
MERCURY   201     80
BISMUTH   209     83
RADON     222     86
THORIUM   232     90
URANIUM   238     92
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ENDLIST

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(Continued)

ZAV	TABLE	INDEX	(INTERPOLATE AGAINST ELECT. TEMP)
ATUM			
HELIUM3	TABLE		136
HELIUM4	TABLE		136
LITHIUM	TABLE		137
BERYLIUM	TABLE		138
BORON	TABLE		139
CARBON	TABLE		86
NITROGEN	TABLE		87
OXYGEN	TABLE		79
FLUORINE	TABLE		140
NEON	TABLE		141
SODIUM	TABLE		142
MAGNESIUM	TABLE		143
ALUMINUM	TABLE		88
SILICON	TABLE		144
SULPHUR	TABLE		145
ARGON	TABLE		146
POTASSIUM	TABLE		147
CALCIUM	TABLE		148
SCANDIUM	TABLE		149
TITANIUM	TABLE		89
VANADIUM	TABLE		150
CHROMIUM	TABLE		151
IRON	TABLE		90
COBALT	TABLE		90
NICKEL	TABLE		91
COPPER	TABLE		92
ZINC	TABLE		152
ARSENIC	TABLE		153
KRYPTON	TABLE		154
RUBIDIUM	TABLE		155
ZIRCONIUM	TABLE		156
NIOBIUM	TABLE		93
MOLYBDENUM	TABLE		81
RHODIUM	TABLE		157
SILVER	TABLE		158
TIN	TABLE		159
XENON	TABLE		160
CESIUM	TABLE		161
BARIUM	TABLE		162
GADOLINIUM	TABLE		163
TANTALUM	TABLE		164
TUNGSTEN	TABLE		94
IRIDIUM	TABLE		165
GOLD	TABLE		95
MERCURY	TABLE		166
BISMUTH	TABLE		167
RADON	TABLE		168
THORIUM	TABLE		169
URANIUM	TABLE		170
ENDLIST			

DECAY FROM	CONSTANTS (/SECOND)	
H2S	H1S	5.000E+03
H2P	H1S	6.265E+08
H2P	H2S	4.699E+38
H3S	H1S	5.000E+03
H3S	H2S	5.000E+03
H3S	H2P	6.313E+06
ENDLIST		

FUSION INCOMER	TARGET	PER REACTION	OUTGOER1	OUTGOER2	ENERGY (J) TO OUTGOER1	ENERGY (J) TO OUTGOER2
D+	T+		HELIUM4	NEUTRON	5.640E-13	2.280E-12
D+	D+		HELIUM3	NEUTRON	1.314E-13	3.925E-13
D+	D+		H+	T+	4.842E-13	1.614E-13
ENDLIST						

(Continued)

POWER ATOM	RADIATN. TABLE INDEX	(INTERPOLATE AGAINST ELECT. TEMP)
H1S	TABLE	40 (TABLE HAS POWER FOR)
H2S	TABLE	40 (EQUILIBRIUM CONCNS)
H3S	TABLE	40 (OF ALL NEUTRALS, SO)
H2P	TABLE	40 (EACH IS POINTED AT IT)
H0	TABLE	40
D0	TABLE	40
T0	TABLE	40
HELIUM3	TABLE	101
HELIUM4	TABLE	101
LITHIUM	TABLE	102
BERYLIUM	TABLE	103
BORON	TABLE	104
CARBON	TABLE	22
NITROGEN	TABLE	23
OXYGEN	TABLE	37
FLUORINE	TABLE	105
NEON	TABLE	106
SODIUM	TABLE	107
MAGNESIUM	TABLE	108
ALUMINUM	TABLE	24
SILICON	TABLE	109
SULPHUR	TABLE	110
ARGON	TABLE	111
POTASSIUM	TABLE	112
CALCIUM	TABLE	113
SCANDIUM	TABLE	114
TITANIUM	TABLE	25
VANADIUM	TABLE	115
CHROMIUM	TABLE	116
IRON	TABLE	30
COBALT	TABLE	26
NICKEL	TABLE	27
COPPER	TABLE	28
ZINC	TABLE	117
ARSENIC	TABLE	118
KRYPTON	TABLE	119
RUBIDIUM	TABLE	120
ZIRCONIUM	TABLE	121
NIOPBIUM	TABLE	29
MOLYBDEN	TABLE	39
RHOBIUM	TABLE	122
SILVER	TABLE	123
TIN	TABLE	124
XENON	TABLE	125
CESIUM	TABLE	126
BARIUM	TABLE	127
GADOLINIUM	TABLE	128
TANTALUM	TABLE	129
TUNGSTEN	TABLE	30
IRIDIUM	TABLE	130
GOLD	TABLE	31
MERCURY	TABLE	131
BISMUTH	TABLE	132
RADON	TABLE	133
THORIUM	TABLE	134
URANIUM	TABLE	135
ENDLIST	-----	

(Continued)

ELASTIC INCOMER	SCATTER TARGET	TABLE INDEX	(INTERPOLATE AGAINST INCOMER TEMP.)
ELECTRON	H+	TABLE	7
ELECTRON	D+	TABLE	7
ELECTRON	H1S	TABLE	13
ELECTRON	H2S	TABLE	13
ELECTRON	H2P	TABLE	13
ELECTRON	H3S	TABLE	13
ELECTRON	HC	TABLE	13
ELECTRON	DG	TABLE	13
ELECTRON	T+	TABLE	7
ELECTRON	TC	TABLE	13
ELECTRON	HELIUM3	TABLE	82
ELECTRON	HELIUM4	TABLE	82
ELECTRON	LITHIUM	TABLE	83
ELECTRON	BERYLIUM	TABLE	84
ELECTRON	BORON	TABLE	85
ELECTRON	CARBON	TABLE	46
ELECTRON	NITROGEN	TABLE	47
ELECTRON	OXYGEN	TABLE	43
ELECTRON	FLUORINE	TABLE	98
ELECTRON	NEON	TABLE	171
ELECTRON	SODIUM	TABLE	172
ELECTRON	MAGNESIUM	TABLE	173
ELECTRON	ALUMINUM	TABLE	48
ELECTRON	SILICON	TABLE	174
ELECTRON	SULPHUR	TABLE	175
ELECTRON	ARGON	TABLE	176
ELECTRON	POTASSIUM	TABLE	177
ELECTRON	CALCIUM	TABLE	178
ELECTRON	SCANDIUM	TABLE	179
ELECTRON	TITANIUM	TABLE	49
ELECTRON	VANADIUM	TABLE	180
ELECTRON	CHROMIUM	TABLE	181
ELECTRON	IRON	TABLE	44
ELECTRON	COBALT	TABLE	50
ELECTRON	NICKEL	TABLE	51
ELECTRON	COPPER	TABLE	52
ELECTRON	ZINC	TABLE	182
ELECTRON	ARSENIC	TABLE	183
ELECTRON	KRYPTON	TABLE	184
ELECTRON	RUBIDIUM	TABLE	185
ELECTRON	ZIRCONIUM	TABLE	186
ELECTRON	NIOBIUM	TABLE	53
ELECTRON	MOLYBDENUM	TABLE	45
ELECTRON	RHODIUM	TABLE	187
ELECTRON	SILVER	TABLE	188
ELECTRON	TIN	TABLE	189
ELECTRON	XENON	TABLE	190
ELECTRON	CESIUM	TABLE	191
ELECTRON	BARIUM	TABLE	192
ELECTRON	GADOLINIUM	TABLE	193
ELECTRON	TANTALUM	TABLE	194
ELECTRON	TUNGSTEN	TABLE	54
ELECTRON	IRIDIUM	TABLE	195
ELECTRON	GOLD	TABLE	63
ELECTRON	MERCURY	TABLE	196
ELECTRON	BISMUTH	TABLE	197
ELECTRON	RADON	TABLE	198
ELECTRON	THORIUM	TABLE	199
ELECTRON	URANIUM	TABLE	200
H+	H1S	TABLE	2
H+	H2S	TABLE	2
H+	H2P	TABLE	2
H+	H3S	TABLE	2
H+	HC	TABLE	5
H+	H+	TABLE	73
H+	D+	TABLE	42
H+	DO	TABLE	55
H+	T+	TABLE	56
H+	HELIUM3	TABLE	57
H1S	H1S	TABLE	32
H2S	H2S	TABLE	32
H2P	H2P	TABLE	32
H3S	H3S	TABLE	32
H1S	H2S	TABLE	32
H1S	H2P	TABLE	32
H1S	H3S	TABLE	32

(Continued)

H2S	H2P	TABLE	32
H2S	H3S	TABLE	32
H2P	H3S	TABLE	32
H+	HELIUM4	TABLE	58
D+	D+	TABLE	1
D+	DO	TABLE	55
D+	T+	TABLE	60
D+	HELIUM3	TABLE	61
D+	HELIUM4	TABLE	62
DO	H1S	TABLE	32
DO	H2S	TABLE	32
DO	H2P	TABLE	32
DO	H3S	TABLE	32
DO	HC	TABLE	32
DO	TC	TABLE	32
DO	T+	TABLE	55
DO	TO	TABLE	32
DO	HELIUM3	TABLE	55
DO	HELIUM4	TABLE	55
TO	H1S	TABLE	32
T+	H2S	TABLE	32
TO	H2P	TABLE	32
TO	H3S	TABLE	32
TO	HC	TABLE	32
T+	T+	TABLE	76
T+	HELIUM3	TABLE	67
T+	HELIUM4	TABLE	68
HELIUM3	HELIUM3	TABLE	77
HELIUM3	HELIUM4	TABLE	70
HELIUM4	HELIUM4	TABLE	78
H+	TC	TABLE	55
D+	HC	TABLE	55
D+	TO	TABLE	55
T+	TC	TABLE	55
T+	DO	TABLE	55
T+	TO	TABLE	55
HC	HELIUM3	TABLE	55
TO	HELIUM3	TABLE	55
TO	HELIUM4	TABLE	55
HC	HELIUM4	TABLE	55
HC	HC	TABLE	32
TO	TC	TABLE	32
D+	H1S	TABLE	55
D+	H2S	TABLE	55
D+	H2P	TABLE	55
D+	H3S	TABLE	55
T+	H1S	TABLE	55
T+	H2S	TABLE	55
T+	H2P	TABLE	55
T+	H3S	TABLE	55
HELIUM3	H1S	TABLE	55
HELIUM3	H2S	TABLE	55
HELIUM3	H2P	TABLE	55
HELIUM3	H3S	TABLE	55
HELIUM4	H1S	TABLE	55
HELIUM4	H2S	TABLE	55
HELIUM4	H2P	TABLE	55
HELIUM4	H3S	TABLE	55
ENDLIST			

(Continued)

INELAST. INCOMER	REACTION TARGET	TABLE INDEX OUTGOFR	(INTERPLATE AGAINST INCOMER TEMP.
ELECTRON	D+	D0	TABLE 34
ELECTRON	DC	D+	TABLE 35
ELECTRON	T+	TC	TABLE 34
ELECTRON	T+	T+	TABLE 35
ELECTRON	H+	H1S	TABLE 8
ELECTRON	H+	H2S	TABLE 9
ELECTRON	H+	H2P	TABLE 9
ELECTRON	H+	H3S	TABLE 11
ELECTRON	H1S	H+	TABLE 12
ELECTRON	H2S	H+	TABLE 17
ELECTRON	H2P	H+	TABLE 17
ELECTRON	H3S	H+	TABLE 20
ELECTRON	H1S	H2S	TABLE 14
ELECTRON	H1S	H2P	TABLE 15
ELECTRON	H1S	H3S	TABLE 16
ELECTRON	H2S	H1S	TABLE 14
ELECTRON	H2S	H2P	TABLE 18
ELECTRON	H2S	H3S	TABLE 19
ELECTRON	H2P	H1S	TABLE 15
ELECTRON	H2P	H2S	TABLE 18
ELECTRON	H2P	H3S	TABLE 19
ELECTRON	H3S	H1S	TABLE 16
ELECTRON	H3S	H2S	TABLE 19
ELECTRON	H3S	H2P	TABLE 19
ELECTRON	H+	H0	TABLE 34
ELECTRON	H+	H+	TABLE 35
H+	H1S	H2S	TABLE 3
H+	H2S	H2P	TABLE 3
H+	H2P	H3S	TABLE 3
H+	H1S	H2P	TABLE 4
H+	H2S	H3S	TABLE 4
H+	H1S	H3S	TABLE 5
H+	H1S	H+	TABLE 64
H+	H2S	H+	TABLE 65
H+	H2P	H+	TABLE 66
H+	H0	H+	TABLE 69
D+	D+	H+	TABLE 96
D+	T+	T+	TABLE 96
D+	D+	HELIUM3	TABLE 96
D+	T+	HELIUM4	TABLE 97

ENDLIST

TABLE 0 HAS 45 ENTRIES
 MULTIPLY BY 1.1675E+07 FOR CONVERSION TO DEGK
 * NUMBERS BELOW ARE KEV ENERGIES

1.000E-02	2.000E-02	3.000E-02	4.000E-02	5.000E-02	6.000E-02	7.000E-02
8.000E-02	9.000E-02	1.000E-01	2.000E-01	3.000E-01	4.000E-01	5.000E-01
6.000E-01	7.000E-01	8.000E-01	9.000E-01	1.000E+00	2.000E+00	3.000E+00
4.000E+00	5.000E+00	6.000E+00	7.000E+00	8.000E+00	9.000E+00	1.000E+01
1.200E+01	1.400E+01	1.600E+01	1.800E+01	2.000E+01	3.000E+01	4.000E+01
5.000E+01	6.000E+01	7.000E+01	8.000E+01	9.000E+01	1.000E+02	2.500E+02
5.000E+02	7.500E+02	1.000E+03				



